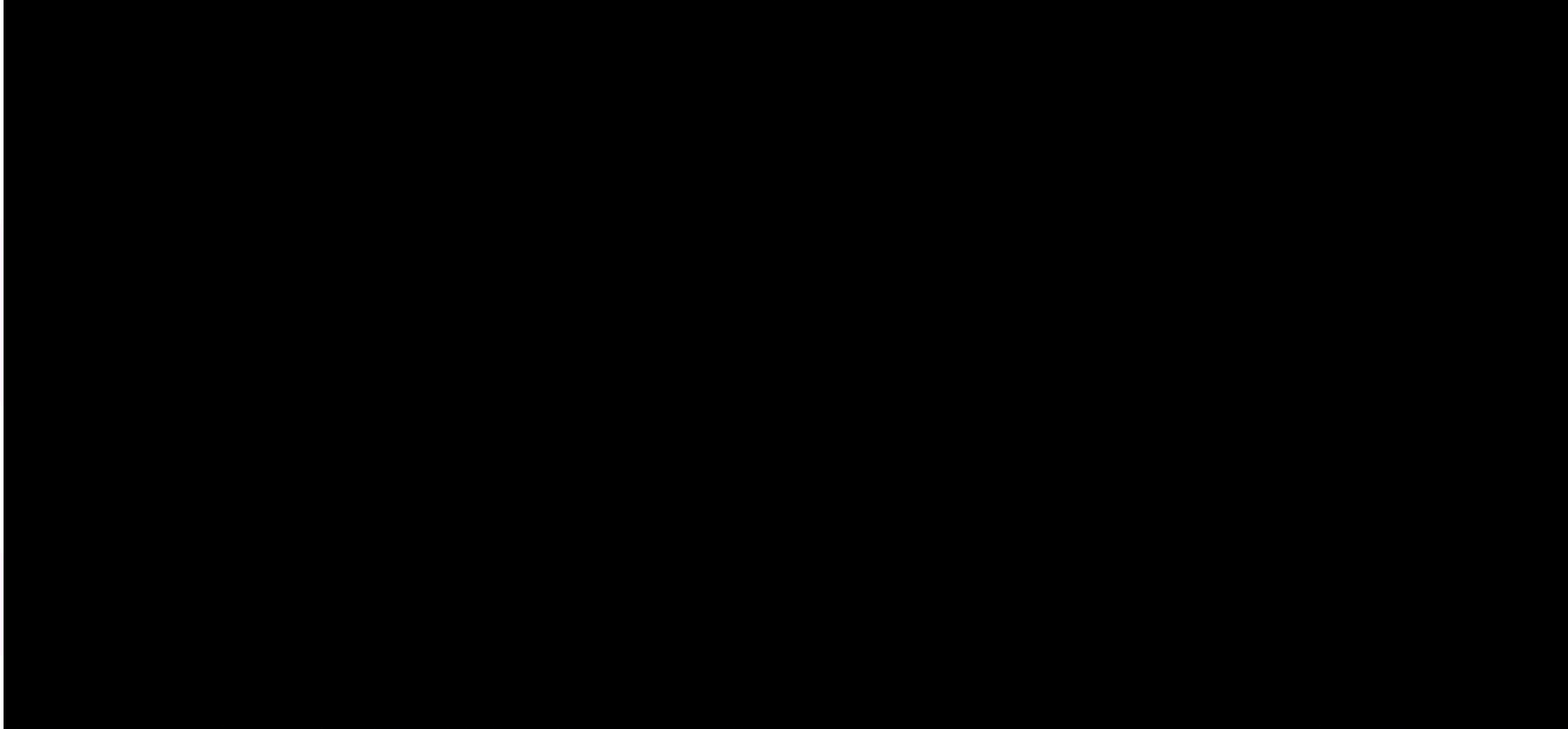


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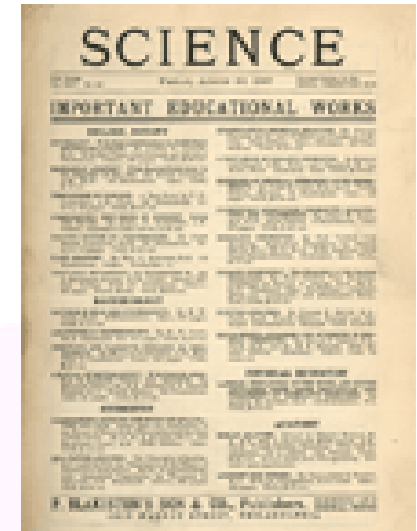
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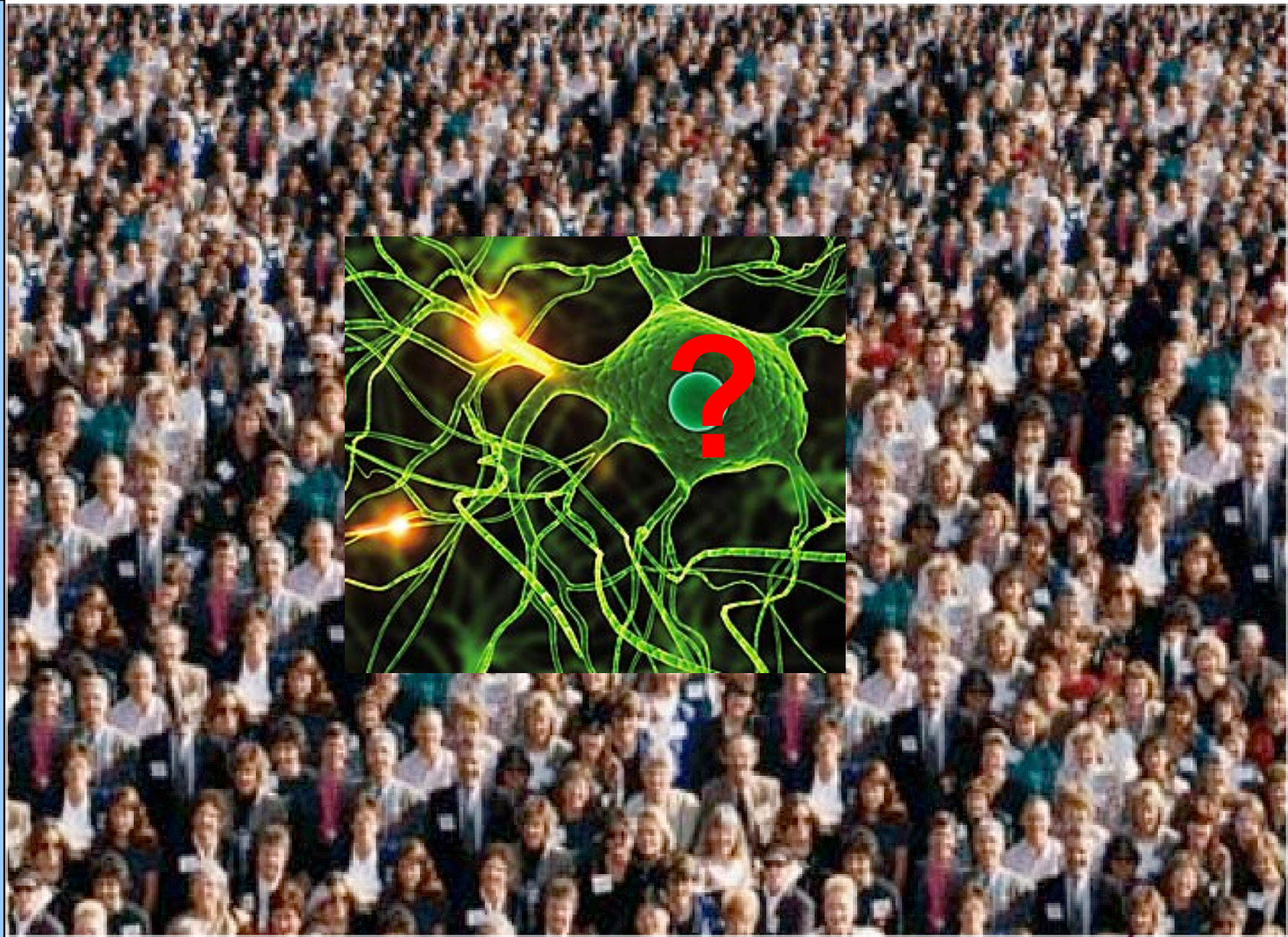


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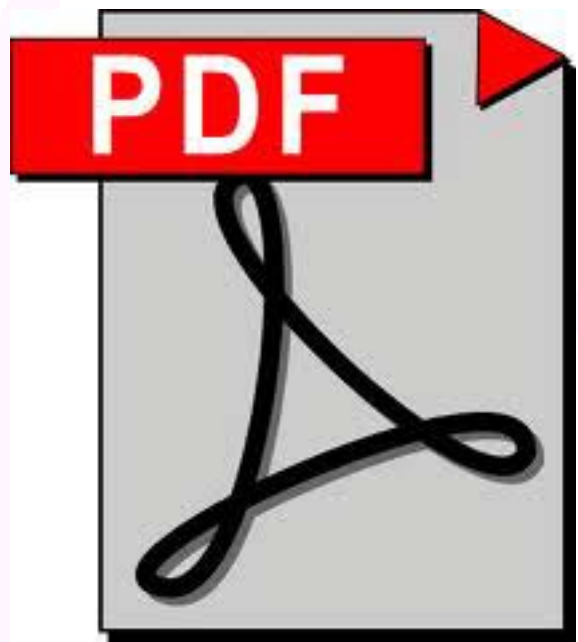


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The image displays two screenshots of the ACS Publications website. The top screenshot shows the journal's homepage with navigation menus, a search bar, and featured articles. The bottom screenshot shows a detailed view of an article titled "Evidence of High •OH Radical Quenching Efficiency by Vitamin B₆".

Article Details:
Title: Evidence of High •OH Radical Quenching Efficiency by Vitamin B₆
Authors: Jon M. Matxain^{1*}, Daniel Padro¹, Mikael Ristola¹, Åke Striffl² and Lief A. Eriksson¹
Affiliations: Kimika Fakultatea, Euskal Herriko Unibertsitatea, P.K. 1072, 20080 Donostia, Euzkadi, Spain, Donostia International Physics Center, 20080 Donostia, Euzkadi, Spain, Bioscience Cooperative Research Centre (Biomagune), Paseo Miramón 182, Ed. Empresarial C, Parque Tecnológico de San Sebastián, 20009, Donostia, Euzkadi, Spain, and Department of Natural Science and Örebro Life Science Center, Örebro University, 70182 Örebro, Sweden
Journal: J. Phys. Chem. B, 2009, 113 (2), pp 9629-9632
DOI: 10.1021/jp903023c
Publication Date (Web): June 26, 2009
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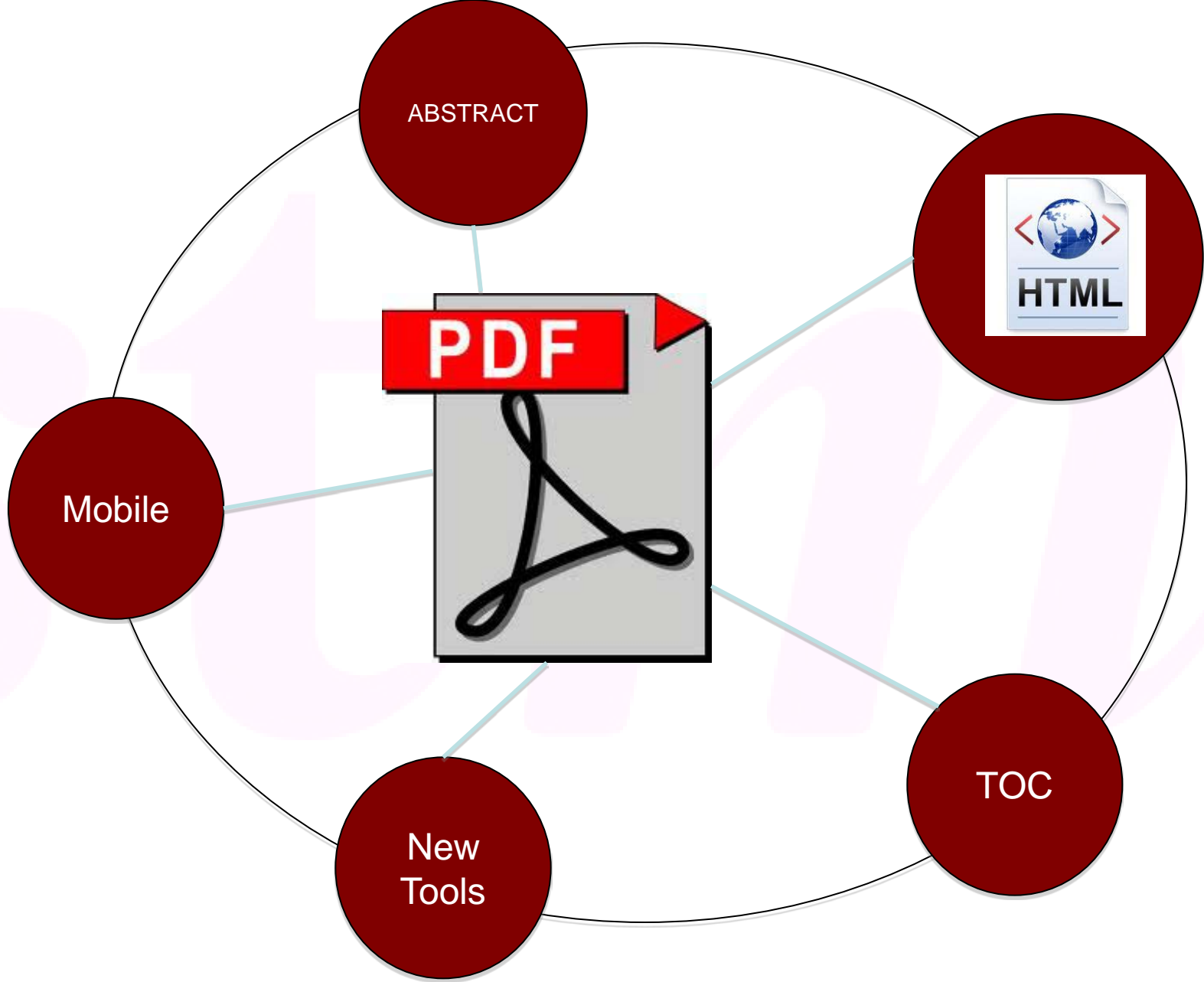
Abstract:
Molecules acting as antioxidants capable of scavenging reactive oxygen species (ROS) are of the utmost importance in the living cell. The antioxidative properties of pyridoxine (vitamin B₆) have recently been discovered. Previous theoretical calculations have shown a high reactivity of pyridoxine toward hydroxyl radicals, where the latter preferably abstract H from either carbon of the two methanol substituents (C8 or C9). In this study, we have explored the reactivity of pyridoxine toward further hydroxyl radicals, considering as the first step

Chemical Reaction:
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Tracking of Multimodal Therapeutic Nanocomplexes Targeting Breast Cancer in Vivo

Rizia Bardhan, Wenxue Chen, Marc Bartels, Carlos Perez-Torres, Maria F. Botero, Robin Ward McAninch, Alejandro Contreras, Rachel Schiff, Robia G. Pautler, Naomi J. Halas, and Amit Joshi
Publication Date (Web): November 22, 2010 (Letter)

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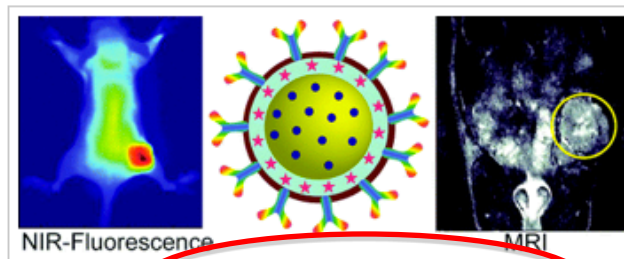


Figure 1 of 7 [Next](#)

Measuring Porosity at the Nanoscale by Quantitative Electron Tomography

E. Biermans, L. Molina, K. J. Batenburg, S. Bals, and G. Van Tendeloo
Publication Date (Web): November 22, 2010 (Letter)
DOI: 10.1021/nl103172r

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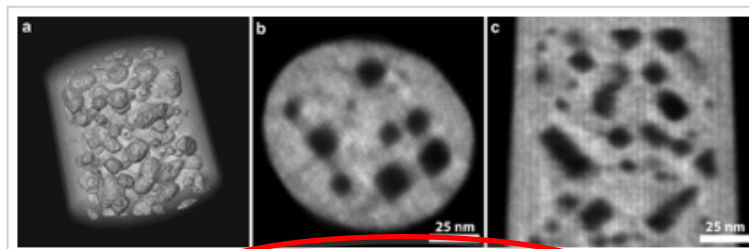


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Simultaneous Noncontact Topography and Electrochemical Imaging by SECM/SICM Featuring Ion Current Feedback Regulation

Yasufumi Takahashi[†], Andrew I. Shevchuk[‡], Pavel Novak[‡], Yumi Murakami[†], Hitoshi Shiku[†], Yuri E. Korchev^{*‡} and Tomokazu Matsue^{*†}

Graduate School of Environmental Studies, Tohoku University, Aramaki Aoba 6-6-11-605, Sendai 980-8579, Japan, and Division of Medicine, **Imperial College** London, Hammersmith Hospital Campus, London W12 0NN, United Kingdom

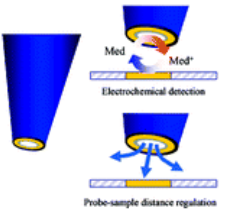
J. Am. Chem. Soc., 2010, 132 (29), pp 10118-10126
 DOI: 10.1021/ja1029478
 Publication Date (Web): June 30, 2010
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matsue@bioinfo.che.tohoku.ac.jp, [†] Tohoku University., [‡] **Imperial College** London.

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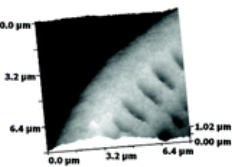
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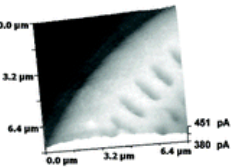


Probe-sample distance regulation

Topography




Electrochemical image



We described a hybrid system of scanning electrochemical microscopy (SECM) and scanning ion conductance microscopy (SICM) with ion current feedback nanopositioning control for simultaneous imaging of noncontact topography and spatial distribution of electrochemical species. A nanopipette/nanoring electrode probe provided submicrometer resolution of the electrochemical measurement on surfaces with complex topology. The SECM/SICM probe had an aperture radius of 720 nm. The inner and outer radii of the SECM Δi nanoring electrode

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Vladimir V. Grushin
Central Research & Development, E. I. DuPont de Nemours & Co., Inc., Experimental Station, Wilmington, Delaware 19880

Acc. Chem. Res., 2010, 43 (1), pp 160-171
DOI: 10.1021/ar9001763
Publication Date (Web): September 29, 2009
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Abstract
Aryl-CF₃ Bond-Forming Reductive Elimination from Palladium(IV)
Nicholas D. Ball, Jeffrey R. Kampf and Melanie S. Sanford
Journal of the American Chemical Society 2010 132 (9), pp 2878-2879
Abstract: This communication describes oxidatively induced Ar-CF₃ bond-forming reductive elimination from new Pd^{IV} complexes of general structure (L-L')Pd^{IV}(Ar)(CF₃). The electrophilic fluorinating reagent, N-fluoro-2,4,6-trimethylpyridinium triflate promotes these ...

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Biography
Vladimir Grushin, a native of Moscow, Russia, obtained his Ph.D. degree from Moscow State University. He then spent several years doing research at the Institute of Organo-Element Compounds

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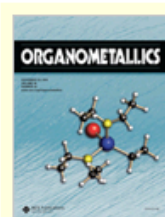
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Jillian A. Hatnean, Robert Beck, Jenna D. Borrelli, and Samuel A. Johnson

Organometallics

2010 29 (22), 6077-6091



Oxidation of a Cyclometalated Pd(II) Dimer with "CF₃⁺": Formation and Reactivity of a Catalytically Competent Monomeric Pd(IV) Aquo Complex

Yingda Ye, Nicholas D. Ball, Jeff W. Kampf, and Melanie S. Sanford

Journal of the American Chemical Society

2010 132 (41), 14682-14687



Fluxionality of [(Ph₃P)₃M(X)] (M = Rh, Ir). The Red and Orange Forms of [(Ph₃P)₃Ir(Cl)]. Which Phosphine Dissociates Faster from Wilkinson's Catalyst?

Jenni Goodman, Vladimir V. Grushin, Roman B. Larichev, Stuart A. Macgregor,

William J. Marshall and D. Christopher Roe

Journal of the American Chemical Society

2010 132 (34), 12013-12026

Carbon-Hydrogen Bond Oxidative Addition of Partially Fluorinated Aromatics to a Ni(PiPr₃)₂ Synthon: The Influence of Steric Bulk on the Thermodynamics and Kinetics of C-H Bond Activation

Jillian A. Hatnean, Robert Beck, Jenna D. Borrelli, and Samuel A. Johnson

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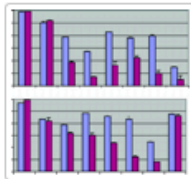
The reaction of (PiPr₃)₂NiCl₂ with the anthracene adduct (THF)₃Mg(η²-C₁₄H₁₀) in THF provides the anthracene adduct (PiPr₃)₂Ni(η²-C₁₄H₁₀). In aromatic solvents (benzene, toluene, mesitylene) a thermal equilibrium exists between the bis(phosphine)nickel(0) ...

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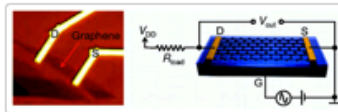
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Pilar Rivera Gil, Günter Oberdorster, Alison Elder, Victor Puentes, and Wolfgang J. Parak



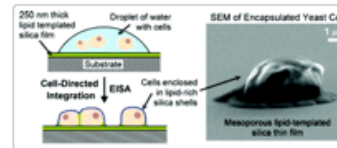
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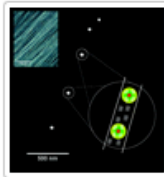
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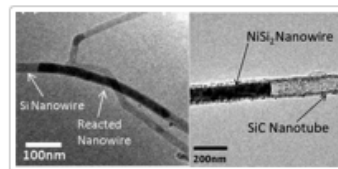
Structure-Property Correlations in Hybrid Polymer-Nanoparticle Electrospun Fibers and Plasmonic Control over their Dichroic Behavior

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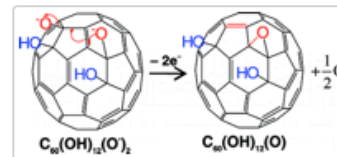
Nanostructural Transformation and Formation of Heterojunctions from Si Nanowires

Tai Lun Wong, Chun Cheng, Wei Li, Kwok Kwong Fung, and Ning Wang



Efficient Synthesis of Fullerene in Anion Form for the Preparation of Electrodeposited Films

Fang F. Wang, Ning Li, Dong Tian, Guo F. Xia, and Ning Xiao



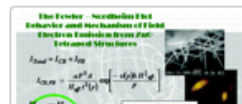
Efficient Inhibition of C-26 Colon Carcinoma by VSMP Gene Delivered by Biodegradable Cationic Nanogel Derived from Polyethyleneimine

MaLing Gou, Ke Men, Juan Zhang, YuHua Li, Jia Song, Shan Luo, HuaShan Shi, YanJun Wen, Gang Guo, MeiJuan Huang, Xia Zhao, ZhiYong Qian, and YuQuan Wei



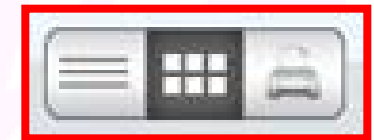
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Ahmed A. Al-Tabbakh, Mahendra A. More, Dilip S. Joag, Imtiaz S. Mulla, and Vijayamohan K. Pillai



Layer-by-Layer Transfer of Multiple, Large Area Sheets of Graphene Grown in Multilayer Stacks on a Single SiC Wafer

Sakulsuk Unarunotai, Justin C. Koepke, Cheng-Lin Tsai, Frank Du, Cesar E. Chialvo, Yuya Murata, Rick Haasch, Ivan Petrov, Nadya Mason, Moonsub Shim, Joseph Lyding, and John A. Rogers



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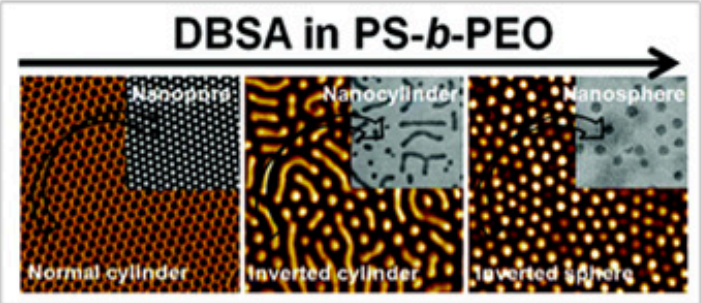
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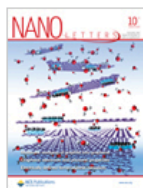
Block Copolymer–Surfactant Complexes in Thin Films for Multiple Usages from Hierarchical Structure to Nano-Objects

Jin Wook Lee, Chansub Lee, Su Yeon Choi and Seung Hyun Kim
DOI: 10.1021/ma901947p

Thin films of block copolymer–surfactant complexes which consist of polystyrene-*b*-poly(ethylene oxide) (PS-*b*-PEO) with dodecylbenzenesulfonic acid (DBSA) hydrogen-bonded to the ether group in PEO blocks, are investigated and characterized by FT-IR, DSC, AFM, TEM, GISAXS, and GIWAXD. The films are solvent-annealed for obtaining controlled orientation and high degree of lateral order of the microdomains. Selective loading of low-mass surfactant into PEO domains leads to morphological transition over wide range of phase diagram in self-assembled nanostructure of block copolymer. In addition, the self-assembly of amphiphilic surfactants combined with block copolymer produces hierarchical structure with multiple length scales and different orientations in thin films. Elimination of surfactants from the complexes by selective solvent, on the other hand, produces various nanostructured materials depending on the amount of surfactants added and microdomain orientation in thin films. Consequently, low-mass amphiphilic surfactants complexed with one component of block copolymer were shown to play multiple roles ranging from generation of hierarchical structures to production of nanoporous films and nano-objects through self-assembly and resultant phase behaviors.



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Phaedon Avouris

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DOI: 10.1021/nl102824h

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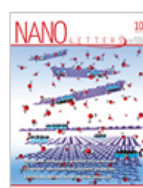
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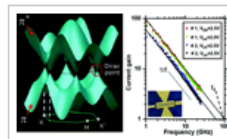
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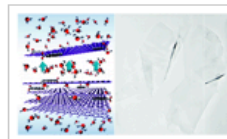
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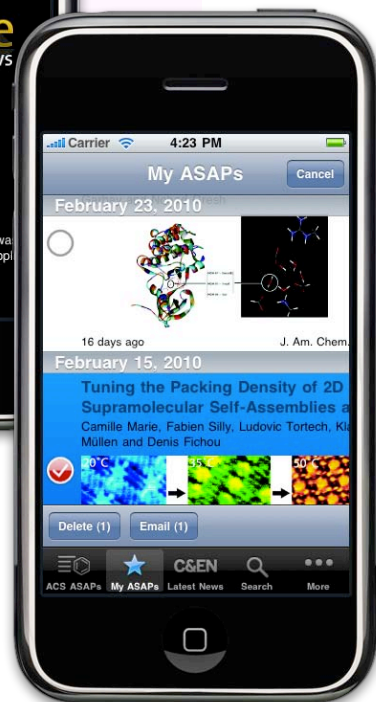
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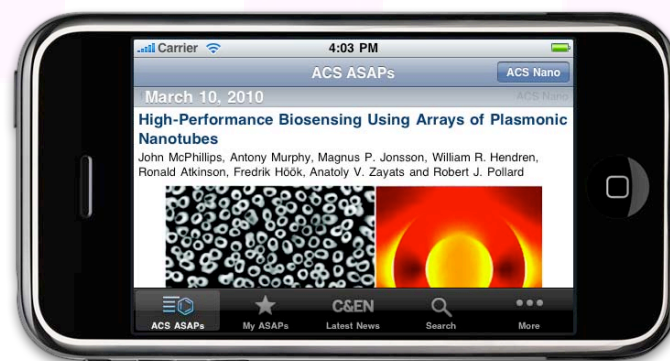


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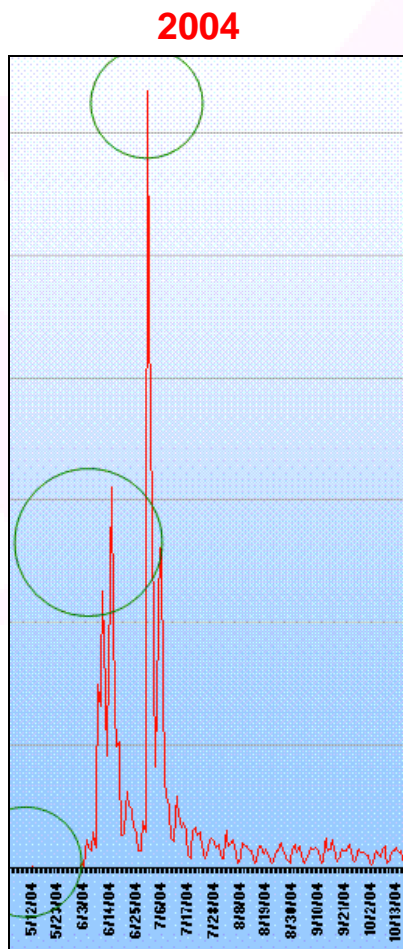
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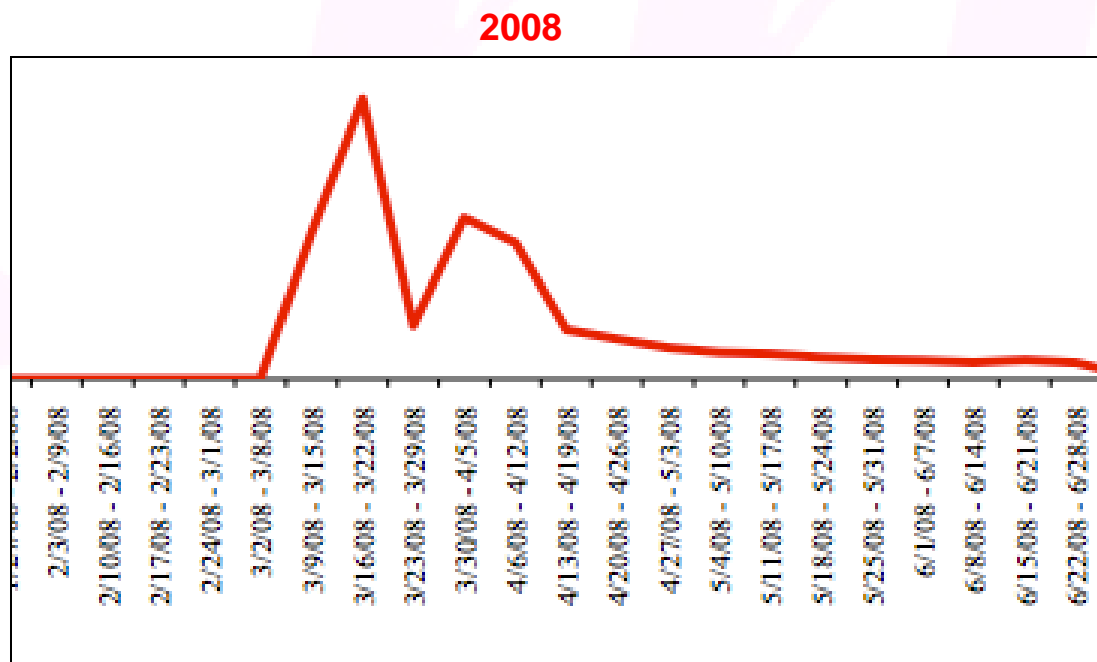
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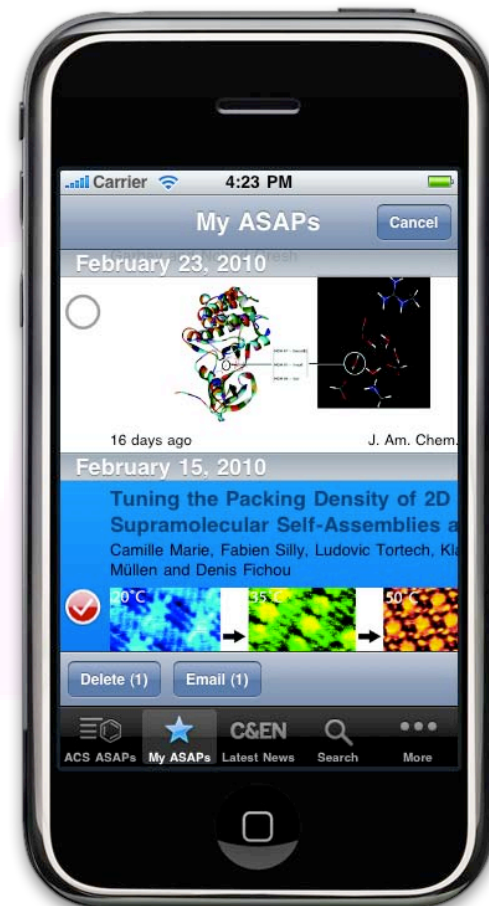
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